Method Development Schemes CHIRAL-AGPTM (conventional detection UV etc)

Characterize your sample and find the starting mobile phase			
Compound type	Starting mobile phase		
Hydrophobic amine	10 mM ammonium or sodium acetate buffer pH 4.5		
Hydrophilic amine Weak acid (phenol etc) Nonprotolyte (amide, ester, alcohol etc)	5% 2-propanol in 10 mM sodium phosphate buffer pH 7.0		
Strong acid (carboxylic acid)	10 mM sodium phosphate buffer pH 7.0		
If the compound contains many functional groups, the protolytic function decides which of the following schemes to use			

Hydrophobic amine Depending on the result obtained with the starting mobile phase, continue as indicated below:					
Retention and enantio- selectivity	No or low enantioselec- tivity and low retention	Enantioselectivity and too high retention.	Too high retention. No enantioselectivity.		
Optimize with pH and/or uncharged modifiers	Increase pH stepwise and adjust retention with 2- propanol (lower conc. gives higher enantioselec- tivity Test another uncharged modifier: acetonitrile, methanol, 1-propanol, ethanol Test low conc. of a charged modifier*: - octanoic acid 1-20 mM - hexanoic or heptanoic acid 1-20 mM - tetraethyl- and tetra propyl-ammonium bromide 1-5 mM	Decrease pH to 4 and/or add 2-propanol Test another uncharged modifier, acetonitrile, methanol, 1-propanol, ethanol	Test different uncharged modifiers: 2-propanol, acetonitrile, methanol, 1- propanol, ethanol Test low conc. of a charged modifier*: - octanoic acid 1-20 mM - hexanoic or heptanoic acid 1-20 mM - tetraethy1- and tetra- propylammonium bro- mide 1-5 mM		

Hydrophilic amines, weak acids and non-protolytesDepending on the result obtained with the starting mobile phase, continue as indicated below:					
Retention and enantio- selectivity	No or low enantioselec- tivity and low retention	Enantioselectivity and too high retention.	Too high retention. No enantioselectivity.		
Optimize with pH and/or uncharged modifiers and/ or buffer conc.	Decrease the 2-propanol conc Test another uncharged modifier: acetonitrile, methanol, 1-propanol, ethanol Amines: Test low conc. of a chargedmodifier*: - octanoic acid 1-20 mM - hexanoic or heptanoic acid 1-20 mM - tetraethyl- and tetra- propylammonium bro mide 1-5 mM		Go to the scheme for hydrophobic amines result obtained with the ase, continue as indicated		
Strong acid (for e	xample carboxylic a	acids) below:	,		
Retention and enantio- selectivity	No or low enantioselec- tivity and low retention	Enantioselectivity and too high retention.	Too high retention. No enantioselectivity.		
Optimize with pH and/or uncharged modifiers and/ or buffer conc	Decrease pH and/or in- crease buffer conc. up to approx. 50-75 mM (max. 100 mM) Try low conc. (1-5 mM) of the charged modifier* DMOA (N,N-dimethyl- octylamine)	Add 2-propanol Test different uncharged modifiers: acetonitrile, methanol, 1-propanol, ethanol	Test different uncharged modifiers: 2-propanol, acetonitrile, methanol, 1-propanol, ethanol Try low conc. (1-5 mM) of the charged modifier* DMOA (N,N-dimethyl- octylamine)		

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*Please note!

If a column has been used with a mobile phase containing a charged modifier, it should not be used afterwards with mobile phases consisting of a pure buffer or a buffer containing an uncharged modifier. Different types of charged modifiers should not be used on the same column.

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